



IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re application of :
Heribert SCHMITT-WILLICH et al. :
Serial No.: 08/319,357 :
Filed: October 6, 1994 :
For: DERIVATIZED DTPA COMPLEXES, PHARMACEUTICAL AGENTS CONTAINING THESE COMPOUNDS, THEIR USE, AND PROCESSES FOR THEIR PRODUCTION :
Group Art Unit: 2203
Examiner: L. Chapman

DECLARATION UNDER 37 C.F.R. §1.132

Honorable Commissioner of
Patents and Trademarks
Washington, D.C. 20231

SIR:

I, Gabriele Schuhmann-Giampieri, being duly warned declare that:

I am a citizen of Germany, residing at Marschnerstrasse 34, 12203 Berlin, Germany.

I possess the degree of Doctor of Natural Sciences, having studied Pharmacy at the Ludwig-Maximilians-Universitaet in Munich.

I am a member of the German Society of Pharmacology and Toxicology.

Since May 1988, I have been employed as a pharmacist by Schering, Aktiengesellschaft, Berlin, Germany, and am presently head of a research group for the pharmacology of contrast media.

Under my supervision, excretion and relaxivity experiments were conducted for the chelate complexes gadolinium ethoxybenzyl DTPA (Gd-EOB) and gadolinium methoxybenzyl DTPA. The structures of these two chelate complexes are shown below:

The chemical structure shows a central gadolinium ion (Gd^{3+}) coordinated by a complex organic ligand. The ligand features a central benzene ring with a methoxy group ($-OCH_3$) at the para position, which is circled in red. The benzene ring is connected via a methylene bridge to a nitrogen atom. This nitrogen is part of a larger structure that includes two carboxylate groups ($-COO^-$) and a carbonyl group ($C=O$). The entire complex is shown with its counterions, Gd^{3+} and O^- .

Chemical structure of the gadolinium complex of the gadoterate meglumine derivative. The structure shows a central Gd^{3+} ion coordinated by three nitrogen atoms of a meglumine backbone and one oxygen atom of a carboxylate group. The meglumine backbone is substituted with a 4-methoxyphenyl group. The carboxylate group is also substituted with a methoxy group.

Using the technique of inductively coupled plasma atomic emission spectrometry (ICP-AES) which is very sensitive for the quantitative measurement of lanthanide ions including gadolinium (detection limit of 65 nmol Gd/L), the following excretion data were obtained in rats (Wistar-Han, 140-160 g).

Relaxivity

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SCH 1412

recovery pulse sequence was used for measuring the T1 relaxation rate. All measurements were conducted at 39-40°C, and at least three increasing concentrations of each test substance were measured beside blank samples.

Additionally, the gadolinium concentration of EOB and of methoxybenzyl-DTPA in the aqueous and in the plasma solution were measured and correlated with measured T1-relaxation rates by using standard least-square algorithm, thus allowing calculation of T1 relaxivity (slope of the function concentration versus relaxation rate). The T1-relaxivity of the methoxybenzyl-DTPA chelate complex was 4.54 L/mmol·sec in water and 6.89 L/mmol·sec in plasma, whereas the T1-relaxivity of the EOB chelate complex was 5.33 L/mmol·sec in water and 8.69 L/mmol·sec in plasma.

The higher biliary excretion of the EOB chelate complex together with the higher efficacy (higher relaxivity) demonstrates unexpected beneficial and advantageous results for using the EOB chelate complex in magnetic resonance imaging of the liver and the biliary system.

I hereby declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under section 1001 of Title 18 of the United States Code, and that such willful false statements may jeopardize the validity of the application or any patent issuing thereon.

April 15, 1996
Date

Gabriele Schuhmann-Giampieri
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